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TABLE OF REDUCED MOBILITY VALUES FROM AMBIENT PRESSURE ION MOBILITY SPECTROMETRY

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1 INTRODUCTION

Since 1970 when ambient pressure ion mobility spectrometry was first introduced as an analytical technique¹ the gas phase mobilities of a wide variety of organic ions have been measured and reported in the literature. Throughout the seventies ion mobility spectrometry (IMS), also called plasma chromatography or gas phase electrophoresis, was the subject of a number of investigative research publications designed to illustrate its potential as a highly sensitive analytical tool, especially useful for trace organic analysis. IMS, however, was never adopted as a routine tool by the general analytical community. Ions under atmospheric conditions could not be separated with the resolution of even a low-resolution mass spectrometer. As the expense of mass spectrometers decreased and their reliability increased at trace levels, the ion mobility spectrometer could no longer offer unique advantages for measurement.

With the development of a unidirectional flow design², inexpensive microprocessor control³, a Fourier transform mode of operation⁴, and a photoionization source⁵, ion mobility has again become a potentially useful tool for analytical chemistry. Rather than a spectrometric technique for purified compounds it now shows promise as a versatile and tunable selective detector for capillary gas chromatography² and capillary supercritical fluid chromatography⁶.

Selective detection of the ion mobility detector is based on selecting an arrival time window which corresponds to the drift time of a particular ion species of interest. This drift time is best determined from a standard at or near the time of analysis. However, it would be convenient to be able to look up potentially interfering compounds in a table.

Since the beginning of atmospheric pressure IMS, mobility data have been reported in terms of reduced mobility constants (K_0) using the simple equation

$$K_0 = \frac{d}{tE} \left(\frac{P}{760} \right) \left(\frac{273}{T} \right)$$

TABLE I
 REDUCED MOBILITY VALUES FROM AMBIENT PRESSURE ION MOBILITY SPECTROMETRY

An asterisk (*) indicates mass identified ions $PI =$ photoionization

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temperature ($^{\circ}C$)	Ref
0.53	Methyl melissate	+		CO ₂ /CO ₂	220	7
0.68	Methyl stearate	+		CO ₂ /CO ₂	220	7
0.76	Methyl myristate	+		CO ₂ /CO ₂	220	7
0.82	Methyl laurate	+		CO ₂ /CO ₂	220	7
0.86	S-(1,2-dicarboxyethyl)cysteine	+	1.00, 1.08, 1.18, 1.26	N ₂ /N ₂	200	8
0.89	Didodecylamine	+		N ₂ /N ₂	148	9
0.89	Methyl caprate	+		CO ₂ /CO ₂	220	7
0.90	L-Cystinyl-bis-L-alanine	+	1.00, 1.08, 1.16	N ₂ /N ₂	200	8
0.90	Lysergic acid diethylamide	+	1.05, 1.16, 1.30, 1.40	N ₂ /N ₂	200	8
0.91	Digitoxigenin	+	1.13, 1.18, 1.42	N ₂ /N ₂	200	8
0.95	<i>n</i> -Octyl acetate	M ₂ H ⁺	1.42	N ₂ /N ₂	136	10
0.97	Methyl caprylate	+		CO ₂ /CO ₂	220	7
0.97	Triethylphosphonoacetate	M ₂ H ⁺	1.23, 1.45	N ₂ /N ₂	150	11
0.98	1-Octanol	+	1.12, 1.21, 1.46, 1.53, 1.88	N ₂ /N ₂	22	12
0.99	Barbital	+	1.50	Air/N ₂	230	13
1.00	L-Cystinyl-bis-L-alanine	+	0.90, 1.08, 1.16	N ₂ /N ₂	200	8
1.00	S-(1,2-Dicarboxyethyl)cysteine	+	0.86, 1.08, 1.18, 1.26	N ₂ /N ₂	200	8
*1.01	Acenaphthalene	(C ₁₂ H ₈) ₃ H ⁺	1.27, 1.84	N ₂ /N ₂	140	14
1.01	<i>n</i> -Heptyl acetate	M ₂ H ⁺	1.50	N ₂ /N ₂	136	10
1.02	Dibutyltinrosamine	[(C ₄ H ₉) ₂ N - NO] ₂ H ⁺	1.56	N ₂ /N ₂	136	15
1.02	Methyl stearate	+		N ₂ /N ₂	150	7
*1.05	Heroin	M ⁺	1.15	N ₂ /N ₂	200	16
1.05	Lysergic acid diethylamide	+	0.90, 1.16, 1.30, 1.40	N ₂ /N ₂	200	8
1.06	1-Octanol	+	1.12, 1.19, 1.45, 1.85	N ₂ /N ₂	100	12
1.06	<i>n</i> -Tetrahydrocannabinol	+		N ₂ /N ₂	200	8
1.07	Dimethyl morpholinophosphoramdate	M ₂ H ⁺	1.14, 1.57, 1.77	N ₂ /N ₂	150	11
1.07	Methyl caproate	+		CO ₂ /CO ₂	220	7
1.08	1-Octanol	+	1.14, 1.22, 1.47, 1.53, 1.83	N ₂ /N ₂	55	12
1.08	Decachlorobiphenyl	C ₁₂ Cl ₁₀ ⁻	1.16, 2.92	N ₂ /N ₂	125	17

1 08	L-Cystinyl-bis-L-alanine	+	0.90, 1.00, 1.16	N ₂ /N ₂	200	8
*1 08	<i>n</i> -Hexyl acetate	M ₂ H ⁺	1.58	N ₂ /N ₂	136	10
1 08	S-(1,2-Dicarboxyethyl)cysteine	+	0.86, 1.00, 1.18, 1.26	N ₂ /N ₂	200	8
1 09	2,4,4'-Di- <i>sec</i> -butylchlorodiphenyl oxide	+		Air/Air	206	18
1.11	Tri- <i>n</i> -hexylamine	+		N ₂ /N ₂	149	9
*1 12	"Tolan"	(C ₁₄ H ₁₀) ₂ H ⁺	1.67	N ₂ /N ₂	140	14
1.12	1-Octanol	+	0.98, 1.21, 1.46, 1.53, 1.88	N ₂ /N ₂	22	12
1.12	1-Octanol	+	1.06, 1.19, 1.45, 1.85	N ₂ /N ₂	100	12
1.12	2,4,2'-Di- <i>sec</i> -Butylchlorodiphenyl oxide	+		Air/Air	206	18
1.13	1-Hexanol	+	1.36, 1.66, 1.74	N ₂ /N ₂	22	12
1.13	Diethyl cyanomethyl phosphonate	M ₂ H ⁺	1.39, 1.48, 1.58, 1.82	N ₂ /N ₂	150	11
1.13	Digitoxigenin	+	0.91, 1.18, 1.42	N ₂ /N ₂	200	8
1.14	1-Octanol	+	1.08, 1.22, 1.47, 1.53, 1.83	N ₂ /N ₂	55	12
1.14	Dimethyl morpholinophosphoramidate	(M ₂ - CH ₃ OCH ₃)H ⁺	1.07, 1.57, 1.77	N ₂ /N ₂	150	11
1.15	Diethyl-2-bromo-5					
*1.15	Heroin	(M - CH ₃ CO ₂) ⁺	1.05	N ₂ /N ₂	200	16
*1.15	Isophthalic acid	[(C ₈ H ₆ O ₄) ₂ - H ₂ O] ⁻	1.58	N ₂ /N ₂	150	19
1.15	Methyl myristate	+		N ₂ /N ₂	150	7
1.15	Trimethylphosphonoacetate	M ₂ H ⁺	1.26, 1.54, 1.68, 1.92	N ₂ /N ₂	150	11
1.16	1-Octanol	+	1.46	N ₂ /N ₂	140	2
*1 16	Cocaine	M ⁺	1.50, 1.84	N ₂ /N ₂	153	16
1 16	Decachlorobiphenyl	C ₁₂ Cl ₁₀ ⁻	1.08, 2.92	N ₂ /N ₂	125	17
1 16	L-Cystinyl-bis-L-alanine	+	0.90, 1.00, 1.08	N ₂ /N ₂	200	8
1 16	Lysergic acid diethylamide	+	0.90, 1.05, 1.30, 1.40	N ₂ /N ₂	200	8
1 16	Octachlorobiphenyl	(C ₁₂ H ₂ Cl ₈)H ⁺		N ₂ /N ₂	125	17
1 16	Octachlorobiphenyl	C ₁₂ H ₂ Cl ₈ ⁻	2.92	N ₂ /N ₂	125	17
1 18	1-Hexanol	+	1.65, 1.81	N ₂ /N ₂	140	2
1 18	1-Iodobutane	+	1.37, 1.57, 1.67, 1.98, 2.2, 2.33	N ₂ /N ₂	135	20
1 18	Digitoxigenin	+	0.91, 1.13, 1.42	N ₂ /N ₂	200	8
1 18	Methyl butyrate	+		CO ₂ /CO ₂	220	7
1.18	<i>n</i> -Pentyl acetate	M ₂ H ⁺	1.85	N ₂ /N ₂	136	10
1 18	S-(1,2-Dicarboxyethyl)cysteine	+	0.86, 1.00, 1.08, 1.26	N ₂ /N ₂	200	8
1.19	1-Octanol	+	1.06, 1.12, 1.45, 1.85	N ₂ /N ₂	100	12
*1 20	7-Methylhexahelicene	(C ₂₇ H ₁₈)H ⁺		N ₂ /N ₂	140	14
1.20	<i>n</i> -Tetradecylamine	+		N ₂ /N ₂	148	9
1.20	Tribenzylamine	+		N ₂ /N ₂	145	9

(Continued on p 144)

TABLE 1 (continued.)

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temperature (°C)	Ref
1.21	1-Octanol	+	0.98, 1.12, 1.46, 1.53, 1.88	N ₂ /N ₂	22	12
1.22	"Endrin"	+		N ₂ /N ₂	124	21
1.22	"Parathion"	M(H ₂ O)H ⁺	1.27, 1.78	N ₂ /N ₂	150	11
1.22	1-Octanol	+	1.08, 1.14, 1.47, 1.53, 1.83	N ₂ /N ₂	55	12
1.22	Hydantoin-5-propionic acid	+	1.38, 1.50	N ₂ /N ₂	200	8
*1.23	Hexahelicene	(C ₂₆ H ₁₆)H ⁺		N ₂ /N ₂	140	14
1.23	Triethylphosphonoacetate	(M ₂ - C ₂ H ₅ OC ₂ H ₅)H ⁺	0.97, 1.45	N ₂ /N ₂	150	11
1.24	Methyl laurate	+		N ₂ /N ₂	150	7
1.24	Methyl propionate	+		CO ₂ /CO ₂	220	7
*1.25	<i>trans</i> -1,4-Distyrylbenzene	(C ₂₂ H ₁₈)H ⁺		N ₂ /N ₂	140	14
1.26	"Fenitrothion"	M(H ₂ O)H ⁺	1.31, 1.40, 1.72, 1.80	N ₂ /N ₂	150	11
*1.26	Butylated hydroxytoluene	(M - H) ⁺		N ₂ /N ₂	210	22
1.26	DDT	+	1.45	N ₂ /N ₂	124	21
1.26	<i>n</i> -Dodecylamine	+		N ₂ /N ₂	148	9
1.26	S-(1,2-Dicarboxyethyl)cysteine	+	0.86, 1.00, 1.08, 1.18	N ₂ /N ₂	200	8
1.26	Trimethylphosphonoacetate	(M ₂ - CH ₃ OCH ₃)H ⁺	1.15, 1.54, 1.68, 1.92	N ₂ /N ₂	150	11
1.27	"Dieldrin"	+		N ₂ /N ₂	124	21
1.27	"Parathion"	MH ⁺	1.22, 1.78	N ₂ /N ₂	150	11
1.27	4,4'- <i>sec</i> -Butylchlorodiphenyl oxide	+		Air/Air	206	18
1.28	2,4'- <i>sec</i> -Butylchlorodiphenyl oxide	+		Air/Air	206	18
1.28	Butabarbital	+	1.35	Air/N ₂	230	13
1.28	Butabarbital	-	1.34	Air/N ₂	230	13
1.28	CO ₂ reactant ion	+	2.00	CO ₂ /CO ₂	220	7
1.28	<i>n</i> -Butyl acetate	M ₂ H ⁺	1.34, 1.40, 1.47, 1.54, 1.62, 1.72, 1.86, 1.93, 2.04	N ₂ /N ₂	136	10
1.28	<i>n</i> -Pentadecane	C ₁₅ H ₃₁ ⁺				
1.30	4,2'- <i>sec</i> -Butylchlorodiphenyl oxide	+		N ₂ /N ₂	135	23
1.30	Amobarbital	-	1.64	Air/Air	206	18
1.30	Ethyl-S-2-diisopropylaminoethylmethylphosphonothiolate	MH ⁺		Air/N ₂	230	13
				N ₂ /N ₂	150	11

1 30	Lysergic acid diethylamide	+	0 90, 1 05, 1 16, 1 40	N ₂ /N ₂	200	8
1 31	"Femtrothon"	MH ⁺	1 26, 1 40, 1 72, 1 80	N ₂ /N ₂	150	11
1 31	"Methyl parathion"	M(H ₂ O)H ⁺	1 36, 1 80	N ₂ /N ₂	150	11
1 31	Pentobarbital	-	1 46, 1 53, 1 59	Air/N ₂	230	13
1 31	Secobarbital	+	1 48, 1 55	Air/N ₂	230	13
1 32	1-Hexanol	+	1 62, 1 76, 2 10	N ₂ /N ₂	100	12
1 32	1-Hexanol	+	1 63, 1 72	N ₂ /N ₂	55	12
1 32	2,2'-sec.-Butyl-chlorodiphenyl oxide	+		Air/Air	206	18
1 32	Amobarbital	+	1 36, 1 53	Air/N ₂	230	13
*1 32	Coronene	(C ₂₄ H ₁₂)H ⁺		N ₂ /N ₂	140	14
1 32	Dimethyl terephthalate	+	1 51, 1 69	N ₂ /N ₂	150	24
1 34	"Disyston"	MH ⁺	2 09	N ₂ /N ₂	150	11
1 34	Butobarbital	-	1 28	Air/N ₂	230	13
1 34	Di-n-hexylamine	+		N ₂ /N ₂	145	9
1 34	n-Pentadecane	C ₁₄ H ₂₉ ⁺	1 28, 1 40, 1 47, 1 54, 1 62, 1 72, 1 86, 1 93, 2 04	N ₂ /N ₂	135	23
1 34	n-Tetradecane	C ₁₄ H ₂₉ ⁺	1 46, 1 54, 1 62, 1 72, 1 85, 1 92, 2 04	N ₂ /N ₂	135	23
1 35	Butobarbital	+	1 28	Air/N ₂	230	13
1 35	n-Decylamine	+		N ₂ /N ₂	149	9
1 36	"Methyl parathion"	MH ⁺	1 31, 1 80	N ₂ /N ₂	150	11
1 36	1-Hexanol	+	1 13, 1 66, 1 74	N ₂ /N ₂	22	12
1 36	Amobarbital	+	1 32, 1 53	Air/N ₂	230	13
1 36	Methyl caprate	+		N ₂ /N ₂	150	7
1 37	"Thumet"	MH ⁺		N ₂ /N ₂	150	11
1 37	1-Butanol	+	1 61, 1 76, 1 87, 1 97	N ₂ /N ₂	22	12
1 37	1-Iodobutane	+	1 18, 1 57, 1 67, 1 98, 2 20, 2 33	N ₂ /N ₂	135	20
1 37	4-Chlorobenzophenone	+		N ₂ /N ₂	25	25
1 38	Barbital	-	1 54	Air/N ₂	230	13
1 38	Hydantoin-5-propionic acid	+	1 22, 1 50	N ₂ /N ₂	200	8
1 38	Pentobarbital	+		Air/N ₂	230	13
*1 39	1,2,5,6-Dibenzanthracene	(C ₁₂ H ₁₄)H ⁺		N ₂ /N ₂	140	14
1 39	Aprobarbital	+	1 56, 1 64, 1 75	Air/N ₂	230	13
1 39	Aprobarbital	-	1 61	Air/N ₂	230	13
1 39	Diethyl cyanomethyl phosphonate	MNO ⁺	1 13, 1 48, 1 58, 1 82	N ₂ /N ₂	150	11
1 39	Lutidine	+		N ₂ /N ₂	85	26
1 40	"Femtrothon"	(M-CH ₂ O)H ⁺	1 26, 1 31, 1 72, 1 80	N ₂ /N ₂	150	11
1 40	1-Chloropentane	+	1 88, 1 99, 2 20, 2 31	N ₂ /N ₂	135	20

(Continued on p 146)

TABLE I (continued)

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temperature (°C)	Ref.
1.40	Benzaldehyde	+	1.84, 1.91, 2.03	N_2/N_2	150	24
1.40	Diethylnitrosamine	$[(C_2H_5)_2N-NO]_2H^+$	1.96	N_2/N_2	136	15
1.40	Dimethyl methylphosphonate	M_2H^+	2.06	N_2/N_2	150	11
1.40	Lysergic acid diethylamide	+	0.90, 1.05, 1.16, 1.30	N_2/N_2	200	8
1.40	<i>n</i> -Pentadecane	$C_{13}H_{27}^+$	1.28, 1.34, 1.47, 1.54, 1.62, 1.72, 1.86, 1.93, 2.04	N_2/N_2	135	23
1.40	<i>n</i> -Tridecane	$C_{13}H_{27}^+$	1.46, 1.54, 1.67, 1.72, 1.82, 1.92, 2.04	N_2/N_2	135	23
1.40	Secobarbital	-	1.46	N_2/N_2	230	13
1.41	2,4,6-Trinitrotoluene	$(M+NO)^+$	1.48	Air/ N_2	193	27
1.41	3,4-Dimethyl benzophenone	+		N_2/N_2	193	27
1.41	<i>n</i> -Propyl acetate	M_2H^+	2.02	N_2/N_2	136	10
1.42	2,4,6-Trinitrotoluene	$(M+NO)^+$	1.49	Air/ N_2	193	27
1.42	Digitoxigenin	+	0.91, 1.13, 1.18	N_2/N_2	200	8
1.42	Mephobarbital	-	1.66	Air/ N_2	230	13
1.42	<i>n</i> -Octyl acetate	MH^+	0.95	N_2/N_2	136	10
1.42	Tri- <i>n</i> -butylamine	+		N_2/N_2	147	9
1.43	Dibenzylamine	+		N_2/N_2	145	9
1.43	Phenobarbital	-	1.63	Air/ N_2	230	13
1.44	4-Aminobenzophenone	+		N_2/N_2	25	
1.44	Dicyclohexylamine	+	1.81	N_2/N_2	149	9
1.44	Mephobarbital	+		N_2/N_2	230	13
1.44	Phenobarbital	+		Air/ N_2	230	13
1.45	1-Octanol	+	1.06, 1.12, 1.19, 1.85	N_2/N_2	100	12
1.45	DDT	+	1.26	N_2/N_2	124	21
1.45	<i>m</i> -Phthalic acid methyl ester	+		Air/ N_2	230	28
*1.45	Perylene	$(C_{20}H_{12})H^+$		N_2/N_2	150	14
1.45	Triethylphosphonoacetate	MH^+	0.97, 1.23	N_2/N_2	150	11
1.46	1-Octanol	+	0.98, 1.12, 1.21, 1.53, 1.88	N_2/N_2	22	12
1.46	1-Octanol	+	1.16	N_2/N_2	140	2
1.46	<i>n</i> -Dodecane	$C_{12}H_{25}^+$	1.54, 1.62, 1.72, 1.83, 1.92, 2.04	N_2/N_2	135	23

1.46	<i>n</i> -Tetradecane	$C_{12}H_{25}^+$	1.34, 1.54, 1.62, 1.72, 1.85, 1.92, 2.04	N_2/N_2	135	23
1.46	<i>n</i> -Tridecane	$C_{12}H_{25}^+$	1.40, 1.54, 1.62, 1.72, 1.82, 1.92, 2.04	N_2/N_2	135	23
1.46	Pentobarbital	-	1.31, 1.53, 1.59	N_2/N_2	230	13
1.46	Secobarbital	-	1.40	N_2/N_2	230	13
1.47	"Dimethoate"	MH ⁺		N_2/N_2	150	11
1.47	1-Octanol	+	1.08, 1.14, 1.22, 1.53, 1.83	N_2/N_2	55	12
1.47	<i>n</i> -Pentadecane	$C_{12}H_{25}^+$	1.28, 1.34, 1.40, 1.54, 1.62, 1.72, 1.86, 1.93, 2.04	N_2/N_2	135	23
1.47	<i>p</i> -Phthalic acid methyl ester	+		N_2/N_2	230	28
1.48	2,4,6-Trinitrotoluene	MH ⁺	1.41	N_2/N_2	193	27
*1.48	3,4-Benzopyrene	$(C_{20}H_{12})H^+$		N_2/N_2	140	14
1.48	Diethyl cyanomethyl phosphonate	$M(H_2OH)^+$	1.13, 1.39, 1.58, 1.82	N_2/N_2	150	11
1.48	<i>m</i> -Phthalic acid methyl ester	-		N_2/N_2	230	28
1.48	Secobarbital	+	1.31, 1.55	N_2/N_2	230	13
1.49	1-Chlorohexane	+	1.72, 1.81, 1.90, 2.06	N_2/N_2	135	20
1.49	2,4,6-Trinitrotoluene	+		N_2/N_2	200	29
*1.49	2,4,6-Trinitrotoluene	M ⁻	1.54, 1.59	N_2/N_2	166	30
*1.49	2,4,6-Trinitrotoluene	M ⁻	1.54, 1.59	N_2/N_2	166	30
1.49	2,4,6-Trinitrotoluene	MH ⁺	1.42	N_2/N_2	193	27
1.50	2,4,6-Trinitrotoluene	M ⁻	1.57	N_2/N_2	193	27
1.50	2,4,6-Trinitrotoluene	M ⁻	1.60	N_2/N_2	193	27
1.50	Barbital	+	0.99	N_2/N_2	230	13
*1.50	Cocaine	$(M - C_8H_5CO_2)^+$	1.16, 1.84	N_2/N_2	153	16
1.50	Hydantoin-5-propionic acid	+	1.22, 1.38	N_2/N_2	200	8
1.50	<i>n</i> -Heptyl acetate	MH ⁺	1.01	N_2/N_2	136	10
1.50	<i>n</i> -Octylamine	+		N_2/N_2	149	9
1.50	<i>p</i> -Phthalic acid methyl ester	-	1.57, 1.77, 1.91	N_2/N_2	230	28
*1.50	Terephthalic acid	$(C_8H_6O_4)NO^+$		N_2/N_2	150	19
*1.51	Benzo[<i>c</i>]phenanthrene	$(C_{18}H_{12})H^+$		N_2/N_2	140	14
1.51	Benzophenone	+		N_2/N_2	25	
1.51	Dibromonitrobenzene	$(C_6H_3Br_2NO_2)H^+$		N_2/N_2	148	31
1.51	Dimethyl terephthalate	+	1.32, 1.69	N_2/N_2	150	24
1.52	1-Bromo-2-chlorobenzene	+		N_2/N_2	220	26
*1.52	Isophthalic acid	$(C_8H_6O_4)NO^+$	1.57, 1.77, 1.91	N_2/N_2	150	19
1.52	Isophthalic acid	+	1.57, 1.76, 1.91, 2.14	N_2/N_2	150	24
1.52	Methyl benzoate	+	1.82, 2.05	N_2/N_2	150	24

(Continued on p. 148)

TABLE I (continued)

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temperature ($^{\circ}\text{C}$)	Ref.
1 52	Methyl caprylate	+		N_2/N_2	150	7
*1 52	Phthalic acid	$(\text{C}_8\text{H}_6\text{O}_4)\text{NO}^+$	1 64, 1 77	N_2/N_2	150	19
1 52	Phthalic acid	+	1 64, 1 77, 2 15	N_2/N_2	150	24
1 52	Terephthalic acid	+	1 57, 1 76, 1 91, 2 14	N_2/N_2	150	24
1 53	1-Bromo-2-chlorobenzene	+		N_2/N_2	175	26
1 53	1-Octanol	+	0 98, 1 12, 1 21, 1 46, 1 88	N_2/N_2	22	12
1 53	1-Octanol	+	1 08, 1 14, 1 22, 1 47, 1 83	N_2/N_2	55	12
1 53	Amobarbital	+	1 32, 1 36	Air/ N_2	230	13
1 53	Dimethyl terephthalate	-		N_2/N_2	150	24
1 53	<i>o</i> -Phthalic acid methyl ester	-		Air/ N_2	230	28
1 53	Pentobarbital	-	1 31, 1 46, 1 59	Air/ N_2	230	13
*1 54	1,2-Benzanthracene	$(\text{C}_{18}\text{H}_{12})\text{H}^+$		N_2/N_2	140	14
1 54	1-Bromo-2-chlorobenzene	+		N_2/N_2	140	26
*1 54	2,4,6-Trinitrotoluene	$(\text{M}-\text{H})^-$	1 49, 1 59	Air/ N_2	166	30
*1 54	2,4,6-Trinitrotoluene	$(\text{M}-\text{H})^-$	1 49, 1 59	N_2/N_2	166	30
1 54	2,4,6-Trinitrotoluene	-	1 49, 1 59	Air/ N_2	200	29
1 54	Barbital	-	1 38	Air/ N_2	230	13
1 54	Fluoranthene	+		N_2/N_2	85	26
1 54	<i>n</i> -Dodecane	$\text{C}_{11}\text{H}_{23}^+$	1 46, 1 62, 1 72, 1 83, 1 92, 2 04	N_2/N_2	135	23
1 54	<i>n</i> -Pentadecane	$\text{C}_{11}\text{H}_{23}^+$	1 28, 1 34, 1 40, 1 47, 1 62, 1 72, 1 86, 1 93, 2 04	N_2/N_2	135	23
1 54	<i>n</i> -Tetradecane	$\text{C}_{11}\text{H}_{23}^+$	1 34, 1 46, 1 62, 1 72, 1 85, 1 92, 2 04	N_2/N_2	135	23
1 54	<i>n</i> -Tridecane	$\text{C}_{11}\text{H}_{23}^+$	1 40, 1 46, 1 62, 1 72, 1 82, 1 92, 2 04	N_2/N_2	135	23
1 54	<i>n</i> -Undecane	$\text{C}_{11}\text{H}_{23}^+$	1 67, 1 71, 1 84, 1 92, 2 04	N_2/N_2	135	23
1 54	<i>o</i> -Chlorodiphenyl oxide	+		N_2/N_2	135	23
1 54	<i>o</i> -Phthalic acid methyl ester	+		Air/Air	206	18
1 54	<i>p</i> -Chlorodiphenyl oxide	+		Air/ N_2	230	28
1 54	Trimethylphosphonoacetate	MNO^+		Air/Air	206	18
1 55	1-Bromo-2-chlorobenzene	+	1 15, 1 26, 1 68, 1 92	N_2/N_2	150	11
				N_2/N_2	220	26

1.55	1-Bromo-2-chlorobenzene	+				N ₂ /N ₂	175	26
1.55	2,4-Dinitrotoluene	MNO ⁺	1.62			N ₂ /N ₂	193	27
*1.55	Chrysene	(C ₁₈ H ₁₂)H ⁺				N ₂ /N ₂	140	14
1.55	Fluoranthene	+				N ₂ /N ₂	175	26
1.55	Fluoranthene	+				N ₂ /N ₂	140	26
1.55	Fluoranthene	+				N ₂ /N ₂	110	26
1.55	Secobarbital	+	1.31, 1.48			N ₂ /N ₂	230	13
1.56	1-Bromo-2-chlorobenzene	+				Air/N ₂	140	26
1.56	1-Bromo-2-chlorobenzene	+				N ₂ /N ₂ (PI)	140	26
1.56	1-Bromo-2-chlorobenzene	+				N ₂ /N ₂	110	26
1.56	2,4-Dinitrotoluene	+				N ₂ /N ₂	85	26
1.56	Aprobarbital	MNO ⁺				Air/N ₂	193	27
1.56	Bromochloronitrobenzene	+	1.39, 1.64, 1.75			Air/N ₂	230	13
1.56	Dibutylnitrosamine	(C ₆ H ₃ BrClNO ₂)H ⁺				N ₂ /N ₂	148	31
1.57	1-Butanol	[(C ₄ H ₉)N-NO]H ⁺	1.02			N ₂ /N ₂	136	15
1.57	1-Iodobutane	+	1.61, 1.73, 1.84, 1.96			N ₂ /N ₂	55	12
1.57	2,4,6-Trinitrotoluene	+	1.18, 1.37, 1.67, 1.98, 2.20, 2.33			N ₂ /N ₂	135	20
1.57	3-Chloropropiophenone	(M-NO ₂)	1.50			Air/N ₂	193	27
1.57	Dimethyl morpholinophosphoramidate	+				N ₂ /N ₂	25	
1.57	Ethyl acetate	MH ⁺	1.07, 1.14, 1.77			N ₂ /N ₂	150	11
1.57	Fluoranthene	M ₂ H ⁺	2.06			N ₂ /N ₂	136	10
1.57	Fluoranthene	+				N ₂ /N ₂	220	26
1.57	Fluoranthene	+				N ₂ /N ₂ (PI)	220	26
1.57	Fluoranthene	+				N ₂ /N ₂	175	26
1.57	Fluoranthene	+				N ₂ /N ₂ (PI)	140	26
1.57	Fluoranthene	+				N ₂ /N ₂ (PI)	85	26
*1.57	Isophthalic acid	(C ₈ H ₆ O ₄)H ⁺	1.52, 1.77, 1.91			N ₂ /N ₂	150	19
1.57	Isophthalic acid	+	1.52, 1.76, 1.91, 2.14			N ₂ /N ₂	150	24
*1.57	Terephthalic acid	(C ₈ H ₆ O ₄)H ⁺	1.52, 1.77, 1.91			N ₂ /N ₂	150	19
1.57	Terephthalic acid	+	1.52, 1.76, 1.91, 2.14			N ₂ /N ₂	150	24
1.58	1-Bromo-2-chlorobenzene	+				N ₂ /N ₂ (PI)	110	26
1.58	Diethyl cyanomethyl phosphonate	MH ⁺	1.13, 1.39, 1.48, 1.82			N ₂ /N ₂	150	11
1.58	Fluoranthene	+				N ₂ /N ₂ (PI)	110	26
*1.58	Isophthalic acid	(C ₈ H ₆ O ₄) ⁻	1.15			N ₂ /N ₂	150	19
1.58	Isophthalic acid	-				N ₂ /N ₂	150	24
1.58	<i>n</i> -Hexyl acetate	MH ⁺	1.08			N ₂ /N ₂	136	10
1.59	1-Bromo-2-chlorobenzene	+				N ₂ /N ₂ (PI)	85	26
1.59	2,4,6-Trinitrotoluene	-	1.49, 1.54			Air/N ₂	200	29
1.59	2,4,6-Trinitrotoluene	-	1.49, 1.54			N ₂ /N ₂	166	30

(Continued on p. 150.)

TABLE I (continued)

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temperature (°C)	Ref.
1.59	2,4,6-Trinitrotoluene	-	1.49, 1.54	Air/N ₂	166	29
1.59	<i>m</i> -Acetotoluidide	-		Air/N ₂	230	28
1.59	<i>n</i> -Heptylamine	+		N ₂ /N ₂	146	9
1.59	Penitobarbital	-	1.31, 1.46, 1.53	Air/N ₂	230	13
1.60	1-Butanol	+	1.63, 1.77, 1.92, 2.02	N ₂ /N ₂	100	12
1.60	2,4,6-Trinitrotoluene	(M-NO ₂) ⁻	1.50	N ₂ /N ₂	193	27
*1.60	Fluoranthene	(C ₁₆ H ₁₀)H ⁺		N ₂ /N ₂	140	14
1.60	<i>m</i> -Acetotoluidide	+		Air/N ₂	230	28
1.60	<i>p</i> -Acetotoluidide	-		Air/N ₂	230	28
1.61	1-Butanol	+	1.37, 1.76, 1.87, 1.97	N ₂ /N ₂	22	12
1.61	1-Butanol	+	1.57, 1.73, 1.84, 1.96	N ₂ /N ₂	55	12
1.61	2,4-Dinitrotoluene	-	1.67	Air/N ₂	200	29
1.61	2,4-Dinitrotoluene	M ⁻	1.67	N ₂ /N ₂	193	27
1.61	2,4-Dinitrotoluene	M ⁻	1.67	N ₂ /N ₂	166	30
1.61	2,4-Dinitrotoluene	M ⁻	1.67	Air/N ₂	166	30
1.61	Aprobarbital	-	1.39	Air/N ₂	230	13
1.61	Dichloronitrobenzene	(C ₆ H ₃ Cl ₂ NO ₂)H ⁺		N ₂ /N ₂	148	31
1.61	Fluorene	+		N ₂ /N ₂	220	26
1.61	Fluorene	+		N ₂ /N ₂	175	26
*1.61	Hexachlorobenzene	C ₆ Cl ₅ O ⁻		N ₂ /N ₂	111	32
1.61	<i>o</i> -Acetotoluidide	-		Air/N ₂	230	28
1.61	<i>p</i> -Acetotoluidide	-		Air/N ₂	230	28
1.62	1-Hexanol	+	1.32, 1.76, 2.10	N ₂ /N ₂	100	12
1.62	2,4-Dinitrotoluene	MH ⁺	1.55	N ₂ /N ₂	193	27
1.62	Iodonitrobenzene	(C ₆ H ₄ INO ₂)H ⁺		N ₂ /N ₂	148	32
1.62	<i>n</i> -Decane	C ₁₀ H ₂₁ ⁺	1.71, 1.92, 2.04	N ₂ /N ₂	135	23
1.62	<i>n</i> -Dodecane	C ₁₀ H ₂₁ ⁺	1.46, 1.54, 1.72, 1.83, 1.92, 2.04	N ₂ /N ₂	135	23
1.62	<i>n</i> -Pentadecane	C ₁₀ H ₂₁ ⁺	1.28, 1.34, 1.40, 1.47, 1.54, 1.72, 1.86, 1.93, 2.04	N ₂ /N ₂	135	23
1.62	<i>n</i> -Tetradecane	C ₁₀ H ₂₁ ⁺	1.34, 1.46, 1.54, 1.72, 1.85, 1.92, 2.04	N ₂ /N ₂	135	23

1 62	<i>n</i> -Tridecane	$C_{10}H_{21}^+$	1,40, 1 46, 1 54, 1 72, 1 82, 1 92, 2 04	N_2/N_2	135	23
1 62	<i>n</i> -Undecane	$C_{10}H_{21}^+$	1 54, 1 72, 1 84, 1 92, 2 04	N_2/N_2	135	23
1 62	<i>o</i> -Acetotoluide	+		Air/ N_2	230	28
1 63	"Tabun"	$M(H_2O)H^+$	1 73, 1 82	N_2/N_2	150	11
1 63	1-Butanol	+	1 60, 1 77, 1 92, 2 02	N_2/N_2	100	12
1 63	1-Chlorooctane	+	1 71, 1 82, 1 97, 2 09	N_3/N_2	135	20
1 63	1-Hexanol	+	1 32, 1 72	N_2/N_2	55	12
*1 63	Benzoic acid	$(C_7H_6O_3H_2O)NO^+$	1 74, 1 82	N_2/N_2	150	19
1 63	Benzoic acid	+	1 74, 1 82, 2 05	N_2/N_2	150	24
1 63	Fluorene	+		N_2/N_2	140	26
*1 63	Hexahydrophthalic anhydride	$(M+1)^+$		N_2/N_2	210	22
1 63	Mephobarbital	+	1 44, 1 81	Air/ N_2	230	13
1 63	<i>n</i> -Hexyl acetate	+		N_2/N_2	140	2
1 63	Phenobarbital	-	1 43	Air/ N_2	230	13
1 64	Amobarbital	-	1 30	Air/ N_2	230	13
1 64	Aprobarbital	+	1 39, 1 56, 1 75	Air/ N_2	230	13
*1 64	Phthalic acid	$(C_8H_4O_3)NO^+$	1 52, 1 77	N_2/N_2	150	19
1 64	Phthalic acid	+	1 52, 1 77, 2 15	N_2/N_2	150	24
1 64	Phthalic anhydride	+	1 52, 1 77, 2 15	N_2/N_2	150	24
1 65	"Soman"	MH ⁺	1 76, 1 92, 2 06	N_2/N_2	150	11
1 65	1-Hexanol	+	1 18, 1 81	N_2/N_2	140	2
1 65	Diphenyloxide	+		Air/Air	206	18
1 65	Isobutyrophenone	+		N_2/N_2	230	25
1 65	<i>m</i> -Toluc acid methyl ester	+		Air/ N_2	230	28
1 66	1-Butanol	+	1 78	N_2/N_2	140	2
1 66	1-Hexanol	+	1 13, 1 36, 1 74	N_2/N_2	22	12
1 66	1-Iodoheptane	+	1 83, 1 95, 2 15	N_2/N_2	135	20
1 66	1-Iodopentane	+	1 86, 2 01, 2 20	N_2/N_2	135	20
1 66	Bromonitrobenzene	$(C_6H_4BrNO_2)H^+$		N_2/N_2	148	31
1 66	Di- <i>n</i> -butylamine	+		N_2/N_2	148	9
1 66	Dibromonitrobenzene	$(C_6H_3BrNO_2)^-$		N_2/N_2	148	31
1 66	<i>m</i> -Toluc acid methyl ester	-		Air/ N_2	230	28
1 66	Mephobarbital	$(C_{14}H_{12})^+$	1 42	Air/ N_2	230	13
*1 66	<i>trans</i> -Stilbene	+		N_2/N_2	140	14
1 66	Tri- <i>n</i> -propylamine	+		N_3/N_2	149	9
*1 67	"Tolan"	$(C_{14}H_{10})H^+$	1 12	N_3/N_2	140	14
1 67	1-Iodobutane	+	1 18, 1 37, 1 57, 1 98, 2 20, 2 33	N_2/N_2	135	20

(Continued on p 152)

TABLE I (continued)

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temperature (°C)	Ref.
1.67	2,4,5-Trichlorotoluene	+		N ₂ /N ₂	155	33
*1.67	2,4-Dinitrotoluene	(M-H) ⁻	1.61	N ₂ /N ₂	166	30
*1.67	2,4-Dinitrotoluene	(M-H) ⁻	1.61	Air/N ₂	166	30
*1.67	2,4-Dinitrotoluene	(M-NO ₂) ⁻	1.61	N ₂ /N ₂	193	27
1.67	2,4-Dinitrotoluene	-	1.61	Air/N ₂	200	29
1.67	Bromonitrobenzene	(C ₆ H ₄ BrNO ₂) ⁻	2.60	N ₂ /N ₂	148	31
1.67	Fluorene	+		N ₂ /N ₂	220	26
1.67	Fluorene	+		N ₂ /N ₂	85	26
1.67	Fluorene	+		N ₂ /N ₂	175	26
1.67	N,N-Diethylaniline	+		N ₂ /N ₂	220	26
*1.67	<i>p</i> -Nitrophenol	MNO ⁺	1.79, 1.91	N ₂ /N ₂	204	34
*1.67	<i>p</i> -Nitrophenol	MNO ⁺	1.80, 1.92	N ₂ /N ₂	204	34
1.67	<i>p</i> -Toluic acid methyl ester	+		Air/N ₂	230	28
1.67	<i>p</i> -Toluic acid methyl ester	-		Air/N ₂	230	28
1.68	1-Bromoheptane	+	1.75, 1.83, 1.95, 2.01, 2.13	N ₂ /N ₂	135	20
1.68	2,4-Dinitrotoluene	(M-NO ₂) ⁻		Air/N ₂	193	27
1.68	2-Methylnaphthalene	+		N ₂ /N ₂ (PI)	175	26
1.68	2-Methylnaphthalene	+		N ₂ /N ₂ (PI)	110	26
1.68	2-Methylnaphthalene	+		N ₂ /N ₂ (PI)	85	26
1.68	Anthracene	+		N ₂ /N ₂ (PI)	110	26
1.68	Diisobutylamine	+		N ₂ /N ₂	148	9
*1.68	Dibenzyl	(C ₁₄ H ₁₄)H ⁺		N ₂ /N ₂	140	14
1.68	Fluorene	+		N ₂ /N ₂	220	26
1.68	Fluorene	+		N ₂ /N ₂	140	26
1.68	Fluorene	+		N ₂ /N ₂ (PI)	110	26
1.68	N,N-Diethylaniline	+		N ₂ /N ₂	85	26
1.68	<i>o</i> -Toluic acid methyl ester	+		Air/N ₂	230	28
1.68	Trimethylphosphonoacetate	MH ⁺	1.15, 1.26, 1.54, 1.92	N ₂ /N ₂	150	11
1.69	2-Methylnaphthalene	+		N ₂ /N ₂	175	26
1.69	2-Methylnaphthalene	+		N ₂ /N ₂	110	26
1.69	Anthracene	+		N ₂ /N ₂	140	32
1.69	Diethyl-2-bromoethylphosphonate	(M-Br)H ⁺	1.15	N ₂ /N ₂	150	11

1 69	Dimethyl terephthalate	+	1.32, 1 51	N ₂ /N ₂	150	24
1 69	Fluorene	+		N ₂ /N ₂ (PI)	175	26
1 69	N,N-Diethylaniline	+		N ₂ /N ₂	175	26
1 69	<i>o</i> -Toluic acid methyl ester	-		Air/N ₂	230	28
1 69	Pyridine	+		N ₂ /N ₂ (PI)	85	26
1 69	Pyridine	+		N ₂ /N ₂	85	26
1 69	Quinoline	+		N ₂ /N ₂ (PI)	110	26
1 70	2-Methylnaphthalene	+		N ₂ /N ₂ (PI)	140	26
1 70	2-Methylnaphthalene	+		N ₂ /N ₂	140	26
1 70	2-Methylnaphthalene	+		N ₂ /N ₂	85	26
1 70	4- <i>tert</i> -Butylpyridine	+		N ₂ /N ₂	147	9
1 70	Anthracene	+		N ₂ /N ₂	110	26
1 70	<i>m</i> -Methyl phenetole	+		Air/N ₂	230	28
1 70	N,N-Diethylaniline	+		N ₂ /N ₂ (PI)	220	26
1 70	N,N-Diethylaniline	+		N ₂ /N ₂ (PI)	85	26
1 70	<i>n</i> -Hexylamine	+		N ₂ /N ₂	146	9
1 70	<i>o</i> -Chloronitrobenzene	+		N ₂ /N ₂	125	17
1 71	"Sarm"	(C ₆ H ₅ ClNO ₂)H ⁺	1.84, 1 95	N ₂ /N ₂	150	11
1 71	1-Chlorooctane	M(H ₂ O) ₂ H ⁺	1.63, 1 82, 1.97, 2 09	N ₂ /N ₂	135	20
1 71	2-Methylnaphthalene	+		N ₂ /N ₂ (PI)	220	26
1 71	Anthracene	+		N ₂ /N ₂ (PI)	85	26
1 71	Anthracene	+		N ₂ /N ₂	85	26
1 71	Dimethylnitrosamine	+		N ₂ /N ₂	136	15
1 71	Fluorene	+	2 21	N ₂ /N ₂ (PI)	85	26
1 71	Methyl caproate	+		N ₂ /N ₂	150	7
1 71	<i>n</i> -Decane	C ₉ H ₁₉ ⁺	1 62, 1 92, 2.04	N ₂ /N ₂	135	23
1 71	<i>n</i> -Undecane	C ₉ H ₁₉ ⁺	1 54, 1 62, 1.84, 1 92, 2.04	N ₂ /N ₂	135	23
1 71	N,N-Diethylaniline	+		N ₂ /N ₂	140	26
1 71	N,N-Diethylaniline	+		N ₂ /N ₂	140	26
1 71	<i>o</i> -Chloronitrobenzene	+		N ₂ /N ₂	125	17
1 71	<i>o</i> -Methyl phenetole	+	1.86, 2.92	Air/N ₂	230	28
1 71	Quinoline	+		N ₂ /N ₂	85	26
1 72	"Fenitrothion"	NO ₂ C ₆ H ₅ (CH ₃)OH ⁺	1 26, 1 31, 1 40, 1 80	N ₂ /N ₂	150	11
1 72	1-Bromooctane	+	1.82, 2 00, 2.20	N ₂ /N ₂	135	20
1 72	1-Chlorohexane	+	1.49, 1.81, 1 90, 2 06	N ₂ /N ₂	135	20
1 72	1-Hexanol	+	1 32, 1.63	N ₂ /N ₂	55	12
1 72	2-Methylnaphthalene	+		N ₂ /N ₂	220	26
1 72	Anthracene	+		N ₂ /N ₂ (PI)	175	26

(Continued on p 154)

TABLE I (continued)

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temp-er-ature ($^{\circ}$ C)	Ref
1.72	Anthracene	+		N_2/N_2	175	26
1.72	Chloronitrobenzene	MH^+		N_2/N_2	148	31
1.72	Fluorene	+		N_2/N_2 (PI)	140	26
1.72	Fluorene	+		N_2/N_2	110	26
1.72	N,N-Diethylaniline	+		N_2/N_2 (PI)	175	26
1.72	n-Dodecane	$C_9H_{19}^+$	1.46, 1.54, 1.62, 1.83, 1.92, 2.04	N_2/N_2	135	23
1.72	n-Nonane	$C_9H_{19}^+$	1.83, 2.06	N_2/N_2	135	23
1.72	n-Pentadecane	$C_9H_{19}^+$	1.28, 1.34, 1.40, 1.47, 1.54, 1.62, 1.86, 1.93, 2.04	N_2/N_2	135	23
1.72	n-Tetradecane	$C_9H_{19}^+$	1.34, 1.46, 1.54, 1.62, 1.85, 1.92, 2.04	N_2/N_2	135	23
1.72	n-Tridecane	$C_9H_{19}^+$	1.40, 1.46, 1.54, 1.62, 1.82, 1.92, 2.04	N_2/N_2	135	23
1.72	p-Methyl phenetole	+		N_2/N_2	135	23
1.73	"Tabun"	$M(H_2O)^+$	1.63, 1.82	Air/ N_2	230	28
1.73	1-Butanol	+	1.57, 1.61, 1.84, 1.96	N_2/N_2	150	11
1.73	1-Chloroheptane	+	1.81, 1.95, 2.11	N_2/N_2	55	12
1.73	1-Iodohexane	+	1.81, 1.89, 2.06	N_2/N_2	135	20
*1.73	α -Naphthyl iodide	MH^+		N_2/N_2	135	20
1.73	Anthracene	+		N_2/N_2 (PI)	140	14
1.73	Dichloronitrobenzene	$(C_6H_3ClNO_2)^-$		N_2/N_2 (PI)	220	26
1.73	N,N-Diethylaniline	+		N_2/N_2	148	31
1.73	N,N-Diethylaniline	+		N_2/N_2 (PI)	140	26
1.73	N,N-Diethylaniline	+		N_2/N_2 (PI)	110	26
1.74	o-Dimethoxy benzene	+		Air/ N_2	230	28
1.74	1-Bromohexane	+	1.85, 1.91, 2.07	N_2/N_2	135	20
1.74	1-Chlorobutane	+	1.83, 1.98, 2.13, 2.38	N_2/N_2	135	20
1.74	1-Hexanol	+	1.13, 1.36, 1.66	N_2/N_2	22	12
1.74	Benzoic acid	+	1.63, 1.82, 2.05	N_2/N_2	150	24
*1.74	Benzoic acid	$(C_7H_5O_2)NO^+$	1.63, 1.82	N_2/N_2	150	19
1.74	Chloronitrobenzene	$(C_6H_4ClNO_2)^-$	2.91	N_2/N_2	148	31
*1.74	m-Mononitrotoluene	M^-	1.81, 2.70	Air/ N_2	166	30

*1.74	<i>m</i> -Mononitrotoluene	M ⁻	2.70	N ₂ /N ₂	166	30
1.74	Mononitrotoluene	-	1.81	Air/N ₂	200	29
1.74	<i>N</i> -Methylaniline	M ⁻		N ₂ /N ₂	85	26
*1.74	<i>o</i> -Mononitrotoluene	M ⁻	1.81, 2.70	Air/N ₂	166	30
*1.74	<i>p</i> -Mononitrotoluene	M ⁻	2.70	N ₂ /N ₂	166	30
*1.74	<i>p</i> -Mononitrotoluene	M ⁻	1.81, 2.70	Air/N ₂	166	30
1.74	Propiophenone	+	2.70	N ₂ /N ₂	166	30
1.74	Pyridine	+		N ₂ /N ₂	110	25
1.74	Pyridine	+		N ₂ /N ₂ (PI)	110	26
1.74	Quinoline	+		N ₂ /N ₂	110	26
1.74	Quinoline	+		N ₂ /N ₂ (PI)	220	26
1.74	<i>tert</i> -Butylbenzene	+	2.01, 2.18, 2.34	N ₂ /N ₂	110	26
1.75	1-Bromoheptane	+	1.68, 1.83, 1.95, 2.01, 2.13	N ₂ /N ₂ (PI)	150	5
1.75	Anthracene	+		N ₂ /N ₂	135	20
1.75	Anthracene	+		N ₂ /N ₂ (PI)	140	26
*1.75	Anthracene	MH ⁺		N ₂ /N ₂	140	26
1.75	Aprobarbital	+	1.39, 1.56, 1.64	N ₂ /N ₂	140	14
1.75	Quinoline	+		Air/N ₂	230	13
1.75	<i>tert</i> -Butylbenzene	+		N ₂ /N ₂ (PI)	140	26
1.76	"Soman"	+	2.03, 2.21	N ₂ /N ₂	150	5
1.76	1-Butanol	(M-CH ₃)H ⁺	1.65, 1.92, 2.06	N ₂ /N ₂	150	11
1.76	1-Hexanol	+	1.37, 1.61, 1.87, 1.97	N ₂ /N ₂	22	12
*1.76	Ethylcellosolve acetate	+	1.32, 1.62, 2.10	N ₂ /N ₂	100	12
1.76	Isophthalic acid	(M+H) ⁺	2.11	N ₂ /N ₂	210	22
1.76	Lutidine	+	1.52, 1.57, 1.91, 2.14	N ₂ /N ₂	150	24
1.76	<i>m</i> -Dimethoxybenzene	+		N ₂ /N ₂ (PI)	110	26
1.76	<i>p</i> -Dimethoxybenzene	+		Air/N ₂	230	28
*1.76	Phenanthrene	MH ⁺		Air/N ₂	230	28
1.76	Quinoline	+		N ₂ /N ₂	140	14
1.76	Terephthalic acid	+	1.52, 1.57, 1.91, 2.14	N ₂ /N ₂	220	26
1.77	1-Butanol	+	1.60, 1.63, 1.92, 2.02	N ₂ /N ₂	150	24
1.77	Anthracene	+		N ₂ /N ₂	100	12
1.77	Dimethyl morpholinophosphoramidate	(M-CH ₃ O)H ⁺	1.07, 1.14, 1.57	N ₂ /N ₂	220	26
*1.77	Isophthalic acid	(C ₈ H ₄ O ₃)H ⁺	1.52, 1.57, 1.91	N ₂ /N ₂	150	11
1.77	<i>N</i> -Methylaniline	+		N ₂ /N ₂	150	19
*1.77	Phthalic acid	(C ₈ H ₄ O ₃)H ⁺		N ₂ /N ₂	140	26
*1.77	Phthalic acid	(C ₈ H ₆ O ₄ -H ₂ O) ⁻	1.52, 1.64	N ₂ /N ₂	150	19
				N ₂ /N ₂	150	19

(Continued on p 156)

TABLE 1 (continued)

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temperature (°C)	Ref.
1.77	Phthalic acid	+	1.52, 1.64, 2.15	N_2/N_2	150	24
1.77	Phthalic acid	-	2.15	N_2/N_2	150	24
1.77	Phthalic anhydride	+	1.52, 1.64, 2.15	N_2/N_2	150	24
1.77	Phthalic anhydride	-		N_2/N_2	150	24
1.77	Quinoline	+		N_2/N_2 (PI)	175	26
*1.77	Terephthalic acid	$(C_8H_4O_3)H^+$	1.52, 1.57, 1.91	N_2/N_2	150	19
1.78	"Parathion"	$NO_2C_8H_8OH^+$	1.22, 1.27	N_2/N_2	150	11
1.78	1-Butanol	+	1.66	N_2/N_2	140	2
1.78	Benzylamine	+		N_2/N_2	149	9
1.78	Quinoline	+		N_2/N_2	140	26
1.79	Aniline	+		N_2/N_2	85	26
1.79	N-Methylamine	+		N_2/N_2	110	26
1.79	<i>p</i> -Nitrophenol	MH^+	1.67, 1.91	N_2/N_2	204	34
1.80	"Fenitrothion"	$NO_2C_6H_4OH^+$	1.26, 1.31, 1.40, 1.72	N_2/N_2	150	11
1.80	"Methyl parathion"	$NO_2C_6H_4OH^+$	1.31, 1.36	N_2/N_2	151	11
1.80	Fluoronitrobenzene	$(C_6H_4FNO_2)H^+$		N_2/N_2	148	31
1.80	Naphthalene	+		N_2/N_2	140	26
*1.80	<i>p</i> -Nitrophenol	MH^+	1.67, 1.92	N_2/N_2	204	34
1.81	1-Chloroheptane	+	1.73, 1.95, 2.11	N_2/N_2	135	20
1.81	1-Chlorohexane	+	1.49, 1.72, 1.90, 2.06	N_2/N_2	135	20
1.81	1-Hexanol	+	1.18, 1.65	N_2/N_2	140	2
1.81	1-Iodobenzene	+	1.73, 1.89, 2.06	N_2/N_2	135	20
1.81	2,4-Dichlorotoluene	+		N_2/N_2	155	33
1.81	Iodobenzene	MH^+		N_2/N_2	125	17
*1.81	<i>m</i> -Mononitrotoluene	$(M-H)^-$	1.74, 2.70	Air/ N_2	166	30
1.81	Mephobarbital	+	1.44, 1.63	Air/ N_2	230	13
1.81	Mononitrotoluene	+	1.74	Air/ N_2	200	29
1.81	<i>n</i> -Bromooctane	$C_8H_{17}^+$		N_2/N_2	135	23
1.81	Naphthalene	+		N_2/N_2 (PI)	140	26
1.81	Naphthalene	+		N_2/N_2	85	26
*1.81	Nitrobenzene	$(C_6H_5NO_2)^-$		N_2/N_2	210	22
*1.81	<i>o</i> -Mononitrotoluene	$(M-H)^-$	1.74, 2.70	Air/ N_2	166	30

*1 81	<i>p</i> -Mononitrotoluene	(M-H) ⁻	1.74, 2.70	Air/N ₂	166	30
1 81	Quinoline	+		N ₂ /N ₂	175	26
1 82	"Tabun"	MH ⁺	1 63, 1.73	N ₂ /N ₂	150	11
1 82	1-Bromoocane	+	1 72, 2 00, 2 20	N ₂ /N ₂	135	20
1 82	1-Chloroocane	+	1 63, 1.71, 1 97, 2 09	N ₂ /N ₂	135	20
1 82	Acetophenone	+		N ₂ /N ₂		25
1 82	Azulene	+		N ₂ /N ₂ (PI)	140	26
1 82	Azulene	+		N ₂ /N ₂	140	26
1 82	Azulene	+		N ₂ /N ₂	110	26
1 82	Benzoic acid	+	1 63, 1 74, 2 05	N ₂ /N ₂	150	24
*1 82	Benzoic acid	(C ₇ H ₆ O ₂)H ⁺	1 63, 1.74	N ₂ /N ₂	150	19
1 82	Diethyl cyanomethyl phosphonate	(M-CN)H ⁺	1.13, 1 39, 1 48, 1 58	N ₂ /N ₂	150	11
1 82	Fluoronitrobenzene	(C ₆ H ₄ FNO ₂) ⁻		N ₂ /N ₂	148	31
1 82	Indoline	+	1.52, 2.05	N ₂ /N ₂ (PI)	85	26
1 82	Methyl benzoate	+		N ₂ /N ₂	150	24
1 82	Methyl benzoate	-		N ₂ /N ₂	150	24
1 83	<i>n</i> -Dodecane	C ₈ H ₁₇ ⁺	1 46, 1.54, 1.62, 1 72, 1 92, 2 04	N ₂ /N ₂	135	23
1 83	<i>n</i> -Nonane	C ₈ H ₁₇ ⁺	1 72, 2 06	N ₂ /N ₂	135	23
1 82	<i>n</i> -Pentylamine	+		N ₂ /N ₂	149	9
1 82	<i>n</i> -Tridecane	C ₈ H ₁₇ ⁺	1 40, 1.46, 1.54, 1.62, 1 72, 1 92, 2 04	N ₂ /N ₂	135	23
1 84	<i>n</i> -Undecane	C ₈ H ₁₇ ⁺	1.54, 1.62, 1 71, 1 92, 2 04	N ₂ /N ₂	110	26
1 82	Naphthalene	+		N ₂ /N ₂	135	23
1 82	Naphthalene	+		N ₂ /N ₂ (PI)	85	26
1 83	1-Bromoheptane	+	1 68, 1 75, 1 95, 2 01, 2 13	N ₂ /N ₂	135	20
1 83	1-Chlorobutane	+	1 74, 1 98, 2 13, 2 38	N ₂ /N ₂	135	20
1 83	1-Iodoheptane	+	1 66, 1 95, 2 15	N ₂ /N ₂	135	20
*1 83	1-Naphthaldehyde	C ₁₁ H ₈ O ⁻		N ₂ /N ₂	111	32
1 83	1-Octanol	+	1 08, 1 14, 1 22, 1 47, 1 53	N ₂ /N ₂	55	12
1 83	Aniline	+		N ₂ /N ₂	140	26
1 83	Aniline	+		N ₂ /N ₂	110	26
1 83	Azulene	+		N ₂ /N ₂ (PI)	110	26
1 83	Azulene	+		N ₂ /N ₂	85	26
1 83	Azulene	+		N ₂ /N ₂	200	35
1 83	<i>n</i> -Amylamine	MH ⁺	2.06	N ₂ /N ₂	135	23
1 83	<i>n</i> -Octane	C ₈ H ₁₇ ⁺	1 71, 1 95	N ₂ /N ₂	150	11
1 84	"Sarin"	M(H ₂ O)H ⁺	1.89, 2 00, 2 15, 2 36	N ₂ /N ₂	150	11
1 84	1-Bromobutane	+		N ₂ /N ₂	135	20

(Continued on p 158)

1.86	Naphthalene	+		N ₂ /N ₂ (PI)	220	26
1.86	Naphthalene	+		N ₂ /N ₂ (PI)	175	26
1.86	<i>o</i> -Chloronitrobenzene	C ₆ H ₄ NO ₂ ⁻		N ₂ /N ₂	125	17
*1.86	<i>p</i> -Nitrophenol	(M-H) ⁻	1.71, 2.92	N ₂ /N ₂	204	34
1.87	1-Butanol	+	1.37, 1.61, 1.76, 1.97	N ₂ /N ₂	22	12
1.87	Indoline	+		N ₂ /N ₂	175	26
1.87	Indoline	+		N ₂ /N ₂	110	26
1.87	<i>m</i> -Cresol	+		N ₂ /N ₂ (PI)	85	26
1.87	<i>m</i> -Xylene	+		N ₂ /N ₂ (PI)	85	26
1.87	Naphthalene	+	2.03, 2.14	N ₂ /N ₂	150	5
1.87	Nitrobenzene	M ⁻		N ₂ /N ₂	125	17
*1.87	<i>p</i> -Nitrobenzaldehyde	C ₇ H ₆ NO ₂ ⁻		N ₂ /N ₂	111	32
*1.87	<i>p</i> -Nitrophenol	(M-H) ⁻		N ₂ /N ₂	204	34
1.88	1-Chloropentane	+	1.40, 1.99, 2.20, 2.31	N ₂ /N ₂	135	20
1.88	1-Octanol	+	0.98, 1.12, 1.21, 1.46, 1.53	N ₂ /N ₂	22	12
1.88	Azulene	+		N ₂ /N ₂ (PI)	175	26
1.88	<i>D</i> - <i>n</i> -propylamine	+		N ₂ /N ₂	149	9
1.88	Indane	+		N ₂ /N ₂ (PI)	110	26
1.88	Indoline	+		N ₂ /N ₂ (PI)	175	26
1.88	Lutidine	+		N ₂ /N ₂	140	26
1.88	Nitrobenzene	MH ⁺		N ₂ /N ₂	125	17
1.89	1-Bromobutane	+	1.84, 2.00, 2.15, 2.36	N ₂ /N ₂	135	20
1.89	1-Iodohexane	+	1.73, 1.81, 2.06	N ₂ /N ₂	135	20
1.89	1-Nitropropane	MH ⁺		N ₂ /N ₂	108	36
1.89	Aniline	+		N ₂ /N ₂	220	26
*1.89	Fumaric acid	M ⁻	2.15	N ₂ /N ₂	150	37
*1.89	Fumaric acid	MH ⁺	1.99	N ₂ /N ₂	150	37
1.89	Indane	+		N ₂ /N ₂	140	26
1.89	Indene	+		N ₂ /N ₂ (PI)	85	26
1.89	Indohne	+		N ₂ /N ₂ (PI)	140	26
1.89	<i>m</i> -Toluidine	+		N ₂ /N ₂ (PI)	85	26
1.89	<i>m</i> -Toluidine	+		N ₂ /N ₂	85	26
1.89	<i>m</i> -Xylene	+		N ₂ /N ₂	85	26
*1.89	Maleic acid	MH ⁺	1.99	N ₂ /N ₂	150	37
1.90	1-Chlorohexane	+	1.49, 1.72, 1.81, 2.06	N ₂ /N ₂	135	20
1.90	Hexanol	M ₂ H ⁺		N ₂ /N ₂	100	36
1.90	Indane	+		N ₂ /N ₂	175	26
1.90	Indene	+		N ₂ /N ₂	110	26

(Continued on p. 160)

TABLE 1 (continued)

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temperature ($^{\circ}\text{C}$)	Ref
1 90	Indoline	+		N_2/N_2 (PI)	220	26
1 90	Indoline	+		N_2/N_2	140	26
1 90	Lutidine	+		N_2/N_2 (PI)	140	26
1 90	Lutidine	+		N_2/N_2 (PI)	110	26
1 90	Lutidine	+		N_2/N_2 (PI)	85	26
1 90	<i>m</i> -Toluidine	+		N_2/N_2	110	26
1 90	N,N-Diethylamine	+		N_2/N_2	220	26
1 90	N-Methylamine	+		N_2/N_2 (PI)	85	26
1 90	Naphthalene	+	2.09, 2.21	N_2/N_2 (PI)	150	5
1 91	1-Bromohexane	+	1.74, 1 85, 2.07	N_2/N_2	135	20
1 91	Aniline	+		N_2/N_2	149	9
1 91	Aniline	+		N_2/N_2	175	26
1 91	Benzaldehyde	+		N_2/N_2	150	24
1 91	Bromobenzene	+		N_2/N_2	125	17
1 91	Diisopropylamine	MH ⁺		N_2/N_2	149	9
1 91	Ethanol	+	2.07	N_2/N_2	55	12
*1 91	Isophthalic acid	(C ₇ H ₅ O)H ⁺	1.52, 1 57, 1 77	N_2/N_2	150	19
1 91	Isophthalic acid	+	1.52, 1 57, 1 76, 2 14	N_2/N_2	150	24
1 91	Lutidine	+		N_2/N_2	220	26
1 91	Lutidine	+		N_2/N_2 (PI)	175	26
1 91	Lutidine	+		N_2/N_2	175	26
1 91	<i>m</i> -Toluidine	+		N_2/N_2	220	26
1 91	<i>m</i> -Toluidine	+		N_2/N_2 (PI)	110	26
1 91	<i>m</i> -Xylene	+		N_2/N_2 (PI)	110	26
1 91	N-Methylamine	+		N_2/N_2	220	26
1 91	N-Methylamine	+		N_2/N_2 (PI)	140	26
*1 91	<i>p</i> -Nitrophenol	(M-31) ⁺	1.67, 1 79	N_2/N_2	204	34
*1 91	Terephthalic acid	(C ₇ H ₅ O)H ⁺	1.52, 1.57, 1 77	N_2/N_2	150	19
1 91	Terephthalic acid	+	1 52, 1.57, 1 76, 2 14	N_2/N_2	150	24
1 92	"Soman"	[M-C(CH ₃) ₂]H ⁺	1 65, 1 76, 2.06	N_2/N_2	150	11
1 92	1-Butanol	+	1.60, 1 63, 1 77, 2.02	N_2/N_2	100	12
1 92	Indane	+		N_2/N_2	220	26

TABLE OF REDUCED MOBILITY VALUES FROM IMS

1.92	Indene	+		N_2/N_2	220	26
1.92	Indene	+		N_2/N_2 (PI)	110	26
1.92	Indene	+		N_2/N_2 (PI)	85	26
1.92	Indoline	+		N_2/N_2	220	26
1.92	Lutidine	+		N_2/N_2 (PI)	220	26
1.92	<i>m</i> -Toluidine	+		N_2/N_2	175	26
1.92	<i>m</i> -Toluidine	+		N_2/N_2 (PI)	140	26
1.92	<i>m</i> -Toluidine	+		N_2/N_2	220	26
1.92	<i>m</i> -Xylene	+		N_2/N_2	110	26
1.92	<i>m</i> -Xylene	+		N_2/N_2	135	23
1.92	<i>n</i> -Bromohexane	+	1.62, 1.71, 2.04	N_2/N_2	135	23
1.92	<i>n</i> -Decane	+	1.46, 1.54, 1.62, 1.72,			
1.92	<i>n</i> -Dodecane	+	1.83, 2.04			
1.92	<i>n</i> -Tetradecane	+	1.34, 1.46, 1.54, 1.62,			
1.92	<i>n</i> -Tridecane	+	1.72, 1.85, 2.04			
1.92	<i>n</i> -Undecane	+	1.40, 1.46, 1.54, 1.62,			
1.92	<i>N,N</i> -Diethylamine	+	1.72, 1.87, 2.04			
1.92	<i>N,N</i> -Diethylamine	+	1.54, 1.62, 1.71, 1.84, 2.04			
1.92	<i>n</i> -Heptane	+				
1.92	<i>p</i> -Nitrophenol	+	2.33			
*1.92	Trimethylphosphonoacetate	+	1.67, 1.80			
1.92	2-Hexanone	+	1.15, 1.26, 1.54, 1.68			
1.93	Ethanol	+	2.10			
1.93	Indane	+				
1.93	Indene	+				
1.93	<i>m</i> -Cresol	+				
1.93	<i>m</i> -Cresol	+				
1.93	<i>m</i> -Xylene	+				
1.93	Methyl butyrate	+				
1.93	<i>n</i> -Pentadecane	+	1.28, 1.34, 1.40, 1.47, 1.54, 1.62,			
1.93	<i>N,N</i> -Diethylamine	+	1.72, 1.86, 2.04			
1.93	<i>N</i> -Methylamine	+				
1.94	1-Bromopentane	+	1.99, 2.07, 2.20			
*1.94	Azulene	$C_{10}H_6^-$			150	7

(Continued on p. 162)

TABLE I (continued)

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temperature (°C)	Ref.
*1.94	Benzaldehyde	$C_7H_6O^-$		N_2/N_2	111	32
1.94	Indene	+		N_2/N_2	175	26
1.94	<i>m</i> -Toluidine	+		N_2/N_2 (PI)	220	26
1.94	<i>m</i> -Toluidine	+		N_2/N_2 (PI)	175	26
1.94	<i>m</i> -Xylene	+		N_2/N_2	175	26
1.94	<i>o</i> -Fluorotoluene	+		Air/ N_2	230	28
1.95	"Sarnin"	MH^+	1.71, 1.84	N_2/N_2	150	11
1.95	1-Bromohexane	+	1.68, 1.75, 1.83, 2.01, 2.13	N_2/N_2	135	20
1.95	1-Chlorohexane	+	1.73, 1.81, 2.11	N_2/N_2	135	20
1.95	1-Iodoheptane	+	1.66, 1.83, 2.15	N_2/N_2	135	20
1.95	2,4-Lutidine	+		Air/ N_2	147	9
1.95	Aniline	+		N_2/N_2	85	26
1.95	Indene	+		N_2/N_2	220	26
1.95	<i>m</i> -Cresol	+		CO_2/CO_2	140	26
1.95	<i>m</i> -Fluorotoluene	+		Air/ N_2	230	28
1.95	<i>m</i> -Xylene	+		N_2/N_2 (PI)	220	26
1.95	<i>m</i> -Xylene	+		N_2/N_2 (PI)	175	26
1.95	<i>m</i> -Xylene	+		N_2/N_2 (PI)	140	26
1.95	<i>p</i> -Fluorotoluene	+		Air/ N_2	230	28
1.95	Phenol	+		N_2/N_2	85	26
1.95	<i>tert</i> -Pentylamine	+		N_2/N_2	149	9
1.95	Triethylamine	+		N_2/N_2	146	9
1.96	1-Butanol	+	1.57, 1.61, 1.73, 1.84	N_2/N_2	55	12
1.96	Diethylnitrosamine	+	1.40	N_2/N_2	136	15
1.96	Indene	+		N_2/N_2 (PI)	175	26
1.96	Indene	+		N_2/N_2 (PI)	140	26
1.96	<i>m</i> -Chlorotoluene	+		N_2/N_2	220	26
1.96	<i>p</i> -Chlorotoluene	+		N_2/N_2	155	33
1.96	Phenol	+		N_2/N_2	85	26
1.97	1-Butanol	+	1.37, 1.61, 1.76, 1.87	N_2/N_2	22	12
1.97	1-Chlorooctane	+	1.63, 1.71, 1.82, 2.09	N_2/N_2	135	20
1.97	<i>m</i> -Chlorotoluene	+		N_2/N_2 (PI)	220	26

1.97	<i>m</i> -Cresol	+		N_2/N_2	230	26
1.97	<i>m</i> -Cresol	+		N_2/N_2	175	26
1.97	<i>m</i> -Cresol	+		N_2/N_2 (PI)	140	26
1.97	<i>n</i> -Butylamine	+		N_2/N_2	146	9
1.98	1-Chlorobutane	+	1.74, 1.83, 2.13, 2.38	N_2/N_2	135	20
1.98	1-Iodobutane	+	1.18, 1.37, 1.57, 1.67, 2.20, 2.33	N_2/N_2	135	20
1.98	Ethanol	+	2.10, 2.17	N_2/N_2	100	12
1.98	<i>m</i> -Chlorotoluene	+		N_2/N_2 (PI)	175	26
1.98	<i>m</i> -Chlorotoluene	+		N_2/N_2	175	26
1.98	<i>m</i> -Xylene	+		N_2/N_2	149	9
1.98	<i>o</i> -Xylene	+		N_2/N_2	149	9
1.98	<i>p</i> -Xylene	+		N_2/N_2	149	9
1.99	1-Bromopentane	+	1.94, 2.07, 2.20	N_2/N_2	135	20
1.99	1-Chloropentane	+	1.40, 1.88, 2.20, 2.31	N_2/N_2	135	20
1.99	Chlorobenzene	MH ⁺		N_2/N_2	125	17
*1.99	Chloropentafluorobenzene	(C ₆ F ₅ O) ⁻		N_2/N_2	111	32
*1.99	Fumaric acid	(M - 18)H ⁺	1.89	N_2/N_2	150	37
1.99	<i>m</i> -Chlorotoluene	+		N_2/N_2	110	26
1.99	<i>m</i> -Chlorotoluene	+		N_2/N_2 (PI)	85	26
1.99	<i>m</i> -Chlorotoluene	+		N_2/N_2	85	26
1.99	<i>m</i> -Cresol	+		N_2/N_2 (PI)	220	26
*1.99	Maleic acid	(M - 18)H ⁺	1.89	N_2/N_2	150	37
2.00	1-Bromobutane	+	1.84, 1.89, 2.15, 2.36	N_2/N_2	135	20
2.00	1-Bromooctane	+	1.72, 1.82, 2.20	N_2/N_2	135	20
2.00	Aniline	+		N_2/N_2	140	26
*2.00	Benzene	(C ₆ H ₆)NO ⁺	2.22	N_2/N_2	50	38
2.00	<i>m</i> -Chlorotoluene	+		N_2/N_2 (PI)	110	26
2.00	<i>m</i> -Cresol	+		N_2/N_2 (PI)	175	26
2.00	<i>n</i> -Butyl acetate	+		N_2/N_2	140	2
2.00	<i>n</i> -Butyl acetate	MH ⁺	1.28	N_2/N_2	136	10
2.00	Phenol	+		N_2/N_2	110	26
2.01	1-Bromoheptane	+	1.68, 1.75, 1.83, 1.95, 2.13	N_2/N_2	135	20
2.01	1-Iodopentane	+	1.66, 1.86, 2.20	N_2/N_2	135	20
2.01	Aniline	+		N_2/N_2	110	26
2.01	Aniline	+		N_2/N_2	85	26
2.01	<i>m</i> -Chlorotoluene	+		N_2/N_2 (PI)	140	26
2.01	<i>m</i> -Chlorotoluene	+		N_2/N_2	140	26
2.01	Phenol	+		N_2/N_2 (PI)	110	26

(Continued on p. 164)

TABLE I (continued)

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temperature ($^{\circ}\text{C}$)	Ref.
2.01	<i>tert</i> -Butylbenzene	+	1.74, 2.18, 2.34	N_2/N_2 (PI)	150	5
2.02	1-Butanol	+	1.60, 1.63, 1.77, 1.92	N_2/N_2	100	12
2.02	<i>n</i> -Propyl acetate	MH^+	1.41	N_2/N_2	136	10
2.02	Pyridine	+		N_2/N_2 (PI)	140	26
2.03	Benzaldehyde	+	1.40, 1.84, 1.91	N_2/N_2	150	24
2.03	Naphthalene	+	1.87, 2.14	N_2/N_2	150	5
2.03	Nitrobenzene	-		N_2/N_2	111	32
2.03	<i>n</i> -Bromohexane	$\text{C}_6\text{H}_{13}^+$		N_2/N_2	135	23
2.03	<i>tert</i> -Butylamine	+		N_2/N_2	148	9
2.03	<i>tert</i> -Butylbenzene	+	1.75, 2.21	N_2/N_2	150	5
2.04	Aniline	+		N_2/N_2 (PI)	110	26
*2.04	Benzene	$(\text{C}_6\text{H}_6)\text{NO}^+$	2.20, 2.23, 2.25	N_2/N_2	93	38
2.04	<i>n</i> -Decane	$\text{C}_6\text{H}_{13}^+$	1.62, 1.71, 1.92	N_2/N_2	135	23
2.04	<i>n</i> -Dodecane	$\text{C}_6\text{H}_{13}^+$	1.46, 1.54, 1.62, 1.72, 1.83, 1.92	N_2/N_2	135	23
2.04	<i>n</i> -Pentadecane	$\text{C}_6\text{H}_{13}^+$	1.28, 1.34, 1.40, 1.47, 1.54, 1.62, 1.72, 1.86, 1.93	N_2/N_2	135	23
2.04	<i>n</i> -Tetradecane	$\text{C}_6\text{H}_{13}^+$	1.34, 1.46, 1.54, 1.62, 1.72, 1.85, 1.92	N_2/N_2	135	23
2.04	<i>n</i> -Tridecane	$\text{C}_6\text{H}_{13}^+$	1.40, 1.46, 1.54, 1.62, 1.72, 1.82, 1.92	N_2/N_2	135	23
2.04	<i>n</i> -Undecane	$\text{C}_6\text{H}_{13}^+$	1.54, 1.62, 1.71, 1.84, 1.92	N_2/N_2	135	23
2.04	<i>o</i> -Xylene	+		N_2/N_2	135	23
2.05	Aniline	+		Air/Air	206	18
*2.05	Azulene	$(\text{C}_{10}\text{H}_8)\text{H}^+$		N_2/N_2	220	26
2.05	Benzoic acid	+	1.63, 1.74, 1.82	N_2/N_2	140	14
2.05	Methyl benzoate	+	1.52, 1.82	N_2/N_2	150	24
2.05	<i>n</i> -Hexane	+		N_2/N_2	150	24
2.05	<i>N</i> -Methylaniline	$\text{C}_6\text{H}_{13}^+$		N_2/N_2	135	23
2.05	Phenol	+		N_2/N_2 (PI)	175	26
2.05	<i>sec</i> -Butylamine	+		N_2/N_2	140	26
2.06	"Soman"	$[\text{M} - \text{C}(\text{CH}_3)_3]\text{H}^+$	1.65, 1.76, 1.92	N_2/N_2	148	9
				N_2/N_2	150	11

2.06	1-Bromohexane	$C_6H_{13}^+$			N_2/N_2	136	36
2.06	1-Chlorohexane	+			N_2/N_2	135	20
2.06	1-Iodohexane	+	1.49, 1.72, 1.81, 1.90		N_2/N_2	135	20
2.06	Aniline	+	1.73, 1.81, 1.89		N_2/N_2	175	26
2.06	Aniline	+			N_2/N_2	140	26
2.06	Dimethyl methylphosphonate	MH^+	1.40		N_2/N_2 (PI)	150	11
2.06	Ethyl acetate	MH^+	1.57		N_2/N_2	136	10
2.06	<i>n</i> -Hexane	$C_6H_{13}^+$			N_2/N_2	136	36
2.06	<i>n</i> -Nonane	$C_6H_{13}^+$	1.72, 1.83		N_2/N_2	135	23
2.06	<i>n</i> -Octane	$C_6H_{13}^+$	1.83		N_2/N_2	135	23
*2.06	Naphthalene	MH^+			N_2/N_2	140	14
2.06	Pyridine	+			N_2/N_2	140	26
2.07	1-Bromohexane	+	1.74, 1.85, 1.91		N_2/N_2	135	20
2.07	1-Bromopentane	+	1.94, 1.99, 2.20		N_2/N_2	135	20
*2.07	Benzene	$(C_6H_6)NO^+$	2.27, 2.32		N_2/N_2	207	38
2.07	Ethanol	+	1.91		N_2/N_2	55	12
2.07	<i>N</i> -Methylaniline	+			N_2/N_2 (PI)	220	26
2.07	Phenol	+			N_2/N_2 (PI)	140	26
*2.08	Water	$(H_2O)_3H^+$	2.13, 2.20, 2.34, 2.67		O_2/Air	160	39
2.08	Aniline	+			N_2/N_2 (PI)	175	26
2.08	Benzene	+	2.33, 2.48		N_2/N_2	150	5
*2.08	Toluene	SF_6^-			Air/SF_6	210	40
2.08	Toluene	+			N_2/N_2	175	26
2.08	Toluene	+			N_2/N_2	110	26
2.09	"Disyston"	+			N_2/N_2 (PI)	85	26
2.09	1-Chlorooctane	$[S(CH_2)_2SCH_3CH_2]H^+$	1.34		N_2/N_2	150	11
2.09	Aniline	+	1.63, 1.71, 1.82, 1.97		N_2/N_2	135	20
2.09	<i>N,N</i> -Diethylaniline	+			N_2/N_2 (PI)	220	26
2.09	Naphthalene	+			N_2/N_2 (PI)	220	26
2.10	1-Hexanol	+	1.90, 2.21		N_2/N_2 (PI)	150	5
2.10	Benzene	+	1.32, 1.62, 1.76		N_2/N_2	100	12
2.10	Ethanol	+	1.93		N_2/N_2	85	26
2.10	Ethanol	+	1.98, 2.17		N_2/N_2	22	12
2.10	Ethanol	+			N_2/N_2	100	12
2.10	<i>N,N</i> -Diethylaniline	+			N_2/N_2 (PI)	175	26
2.10	<i>N,N</i> -Diethylaniline	+			N_2/N_2 (PI)	140	26
2.10	<i>N,N</i> -Diethylaniline	+			N_2/N_2 (PI)	140	26
2.10	Toluene	+			N_2/N_2	140	26

(Continued on p. 166)

TABLE I (continued)

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temperature ($^{\circ}\text{C}$)	Ref
2,10	Toluene	+		N_2/N_2	85	26
2,11	1-Chloroheptane	+	173, 181, 195	N_2/N_2	135	26
2,11	Benzene	+		N_2/N_2 (PI)	85	26
*2,11	Ethylcellulosolve acetate	$(\text{C}_4\text{H}_7\text{O}_2)^+$		N_2/N_2	210	22
2,11	Fluorobenzene	MH^+	176	N_2/N_2	125	17
*2,11	Mercury	M^+		N_2/N_2	140	14
2,11	Phenol	+		N_2/N_2	175	26
2,11	Toluene	+		N_2/N_2	149	9
2,12	Cyclohexane	+		N_2/N_2	149	9
2,12	Diethylamine	+		N_2/N_2	149	9
2,12	Toluene	+		N_2/N_2	149	9
2,13	1-Bromoheptane	+	168, 175, 183, 195, 201	N_2/N_2 (PI)	110	26
2,13	1-Chlorobutane	+	174, 183, 198, 238	N_2/N_2	135	20
2,13	Water	$(\text{H}_2\text{O})_4\text{H}^+$	208, 220, 234, 267	O_2/Air	135	20
*2,13	<i>n</i> -Propylamine	+		N_2/N_2	160	39
2,13	Phenol	+		N_2/N_2	149	9
2,13	Phenol	+		N_2/N_2	220	26
2,13	Phenol	+		N_2/N_2 (PI)	175	26
2,13	Phenol	+		N_2/N_2	85	26
2,13	Toluene	+		N_2/N_2 (PI)	175	26
2,14	Isophthalic acid	+	175, 26	N_2/N_2	150	24
2,14	Naphthalene	+	152, 157, 176, 191	N_2/N_2	150	24
2,14	Phenol	+	187, 203	N_2/N_2	150	5
2,14	Terephthalic acid	+		N_2/N_2 (PI)	220	26
2,15	1-Bromobutane	+	152, 157, 176, 191	N_2/N_2	150	24
2,15	1-Iodoheptane	+	184, 189, 200, 236	N_2/N_2	135	20
*2,15	Fumaric acid	$(\text{M}-18)^-$	166, 183, 195	N_2/N_2	135	20
*2,15	Maleic acid	$(\text{M}-18)^-$	189	N_2/N_2	150	37
2,15	Phthalic acid	+	152, 164, 177	N_2/N_2	150	37
2,15	Phthalic acid	-	177	N_2/N_2	150	24
2,15	Phthalic anhydride	+	152, 164, 177	N_2/N_2	150	24
2,15	Toluene	+		N_2/N_2 (PI)	140	26
2,16	<i>n</i> -Bromopentane	$\text{C}_5\text{H}_{11}^+$		N_2/N_2	135	23

2.16	<i>n</i> -Propylamine	MH ⁺	1.98, 2.10	N ₂ /N ₂	200	35
2.16	Pyridine	+		N ₂ /N ₂ (PI)	220	26
2.16	Pyridine	+		N ₂ /N ₂	220	26
2.16	Pyridine	+		N ₂ /N ₂ (PI)	175	26
2.16	Pyridine	+		N ₂ /N ₂	175	26
2.16	Toluene	+		N ₂ /N ₂	220	26
2.17	Ethanol	+		N ₂ /N ₂	100	12
2.17	Toluene	+		N ₂ /N ₂ (PI)	220	26
2.18	Benzene	+		N ₂ /N ₂	110	26
2.18	<i>n</i> -Pentane	C ₅ H ₁₁ ⁺		N ₂ /N ₂	135	23
2.18	<i>tert</i> -Butylbenzene	+	1.74, 2.01, 2.34	N ₂ /N ₂ (PI)	150	5
2.19	Isopropylamine	+		N ₂ /N ₂	149	9
2.19	Pyridine	+		N ₂ /N ₂	145	9
2.20	<i>i</i> -Bromooctane	+	1.72, 1.82, 2.00	N ₂ /N ₂	135	20
2.20	1-Bromopentane	+	1.94, 1.99, 2.07	N ₂ /N ₂	135	20
2.20	1-Chloropentane	+	1.40, 1.88, 1.99, 2.31	N ₂ /N ₂	135	20
2.20	1-Iodobutane	+	1.18, 1.37, 1.57, 1.67, 1.98, 2.33	N ₂ /N ₂	135	20
2.20	1-Iodopentane	+	1.66, 1.86, 2.01	N ₂ /N ₂	135	20
*2.20	Benzene	+	2.04, 2.23, 2.25	N ₂ /N ₂	93	38
2.20	Water	(H ₂ O) ₃ H ⁺	2.08, 2.13, 2.34, 2.67	O ₂ /Air	160	39
2.21	Benzene	+		N ₂ /N ₂ (PI)	110	26
2.21	Dimethylnitrosamine	[(CH ₃) ₂ N-NO]H ⁺	1.71	N ₂ /N ₂	136	15
2.21	Naphthalene	+	1.90, 2.09	N ₂ /N ₂ (PI)	150	5
2.21	<i>tert</i> -Butylbenzene	+	1.75, 2.03	N ₂ /N ₂	150	5
*2.22	Benzene	(C ₆ H ₆) ⁺	2.00	N ₂ /N ₂	150	5
*2.22	Benzene	+	2.00	N ₂ /N ₂	50	38
*2.22	Benzene	[(C ₆ H ₆)(H ₂ O)(N ₂)]H ⁺	2.00	N ₂ /N ₂	50	38
*2.22	Benzene	[(C ₆ H ₆)(N ₂) ₂] ⁺	2.00	N ₂ /N ₂	50	38
*2.22	Benzene	[(C ₆ H ₆)(N ₂)] ⁺	2.00	N ₂ /N ₂	50	38
*2.22	Benzene	[(C ₆ H ₆)(N ₂)H] ⁺	2.00	N ₂ /N ₂	50	38
*2.22	Benzene	[(C ₆ H ₆)(N ₂)N] ⁺	2.00	N ₂ /N ₂	50	38
*2.23	Benzene	(C ₆ H ₆)H ⁺	2.04, 2.20, 2.25	N ₂ /N ₂	50	38
*2.23	Benzene	+	2.04, 2.20, 2.25	N ₂ /N ₂	93	38
*2.23	Benzene	+	2.04, 2.20, 2.25	N ₂ /N ₂	93	38
*2.23	Benzene	[(C ₆ H ₆)(H ₂ O)(N ₂)]H ⁺	2.04, 2.20, 2.25	N ₂ /N ₂	93	38
*2.23	Benzene	[(C ₆ H ₆)(N ₂)] ⁺	2.04, 2.20, 2.25	N ₂ /N ₂	93	38
*2.23	Benzene	[(C ₆ H ₆)(N ₂)H] ⁺	2.04, 2.20, 2.25	N ₂ /N ₂	93	38
2.24	Benzene	+	2.04, 2.20, 2.25	N ₂ /N ₂	140	26

(Continued on p. 168)

TABLE 1 (continued)

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temperature ($^{\circ}\text{C}$)	Ref
2.24	Pyrrrole	+		N_2/N_2	85	26
*2.25	Benzene	$(\text{C}_6\text{H}_6)^+$	2.04, 2.20, 2.23	N_2/N_2	93	38
2.25	Pyrrrole	+		N_2/N_2	110	26
2.26	Benzene	+		N_2/N_2	220	26
2.26	Benzene	+		N_2/N_2 (PI)	140	26
2.27	Water	$(\text{H}_2\text{O})_3\text{H}^+$	2.79	N_2/N_2	136	15
*2.27	Benzene	$(\text{C}_6\text{H}_6)\text{H}^+$	2.07, 2.32	N_2/N_2	207	38
2.27	Benzene	+		N_2/N_2	149	9
2.27	Benzene	+		N_2/N_2	175	26
2.27	Benzene	+		N_2/N_2 (PI)	175	26
2.28	CO_2	+		N_2/N_2	150	7
2.28	Pyrrrole	+		N_2/N_2 (PI)	85	26
2.29	Benzene	+		N_2/N_2 (PI)	220	26
2.29	Pyrrrole	+	1.40, 1.88, 1.99, 2.20	N_2/N_2 (PI)	110	26
2.31	1-Chloropentane	+		N_2/N_2	135	20
2.31	Pyrrrole	+		N_2/N_2	175	26
2.31	Pyrrrole	+		N_2/N_2 (PI)	140	26
2.31	Pyrrrole	+		N_2/N_2	140	26
*2.32	Benzene	$(\text{C}_6\text{H}_6)^+$	2.07, 2.27	N_2/N_2	207	38
2.32	<i>n</i> -Bromobutane	C_4H_9^+		N_2/N_2	135	23
*2.32	SF_6	SF_5^-		N_2/N_2	111	32
2.33	1-Iodobutane	+	1.18, 1.37, 1.57, 1.67, 1.98, 2.20	N_2/N_2	135	20
2.33	Benzene	+	2.08, 2.48	N_2/N_2	150	5
2.33	<i>n</i> -Heptane	C_4H_9^+	1.92	N_2/N_2	135	23
2.34	Water	$(\text{H}_2\text{O})_2\text{H}^+$	2.08, 2.13, 2.20, 2.67	O_2/Air	160	39
2.34	Pyrrrole	+		N_2/N_2	220	26
2.34	<i>tert</i> -Butylbenzene	+	1.74, 2.01, 2.18	N_2/N_2 (PI)	175	26
2.34	Trimethylamine	+		N_2/N_2 (PI)	150	5
2.36	1-Bromobutane	+	1.84, 1.89, 2.00, 2.15	N_2/N_2	149	9
2.38	1-Chlorobutane	+	1.74, 1.83, 1.98, 2.13	N_2/N_2	135	20
2.38	Pyrrrole	+		N_2/N_2 (PI)	135	20
2.38	Pyrrrole	+		N_2/N_2 (PI)	220	26

2.39	Ethylamine	+		N ₂ /N ₂	149	9
*2.39	SF ₆ ⁻	SF ₆ ⁻		N ₂ /N ₂	111	32
2.40	Dimethylamine	+		N ₂ /N ₂	148	9
2.41	CO ₄ ⁻			Air/N ₂	166	41
2.45	Benzene	+	2.61	N ₂ /N ₂ (PI)	150	5
2.46	n-Ethylamine	MH ⁺		N ₂ /N ₂	200	35
*2.47	Benzene	MH ⁺		N ₂ /N ₂	140	14
2.48	Benzene	+	2.08, 2.33	N ₂ /N ₂	150	5
2.50	Iodobenzene	127I ⁻		N ₂ /N ₂	136	42
2.50	All iodinated alkanes	I ⁻		N ₂ /N ₂	135	20
2.51	Iodobenzene	I ⁻		N ₂ /N ₂	125	17
2.51	Iodonitrobenzene	I ⁻		N ₂ /N ₂	148	31
2.53	Iodobenzene	I ⁻		N ₂ /N ₂	140	33
*2.53	Methylamine	+		N ₂ /N ₂	149	9
*2.55	Air	CO ₃ ⁻		Air/Air	210	40
*2.57	Air	CO ₃ ⁻		Air/N ₂	220	43
*2.57	Air	CO ₃ ⁻		Air/N ₂	210	44
*2.58	Air	CO ₃ ⁻		Air/SF ₆	210	40
2.59	Water	(H ₂ O) ₂ O ₂ ⁻	2.94	Air/N ₂	166	41
2.59	Water	(H ₂ O) ₂ O ₂ ⁻	2.94	Air/N ₂	166	41
2.59	All brominated alkanes	Br ⁻		N ₂ /N ₂	135	20
2.59	Bromobenzene	Br ⁻		N ₂ /N ₂	136	42
2.60	Bromonitrobenzene	Br ⁻		N ₂ /N ₂	148	31
2.61	Benzene	+	1.67	N ₂ /N ₂ (PI)	150	5
2.61	Bromobenzene	Br ⁻	2.45	N ₂ /N ₂	125	17
2.63	Bromobenzene	Br ⁻		N ₂ /N ₂	140	33
2.63	n-Methylamine	MH ⁺		N ₂ /N ₂	200	35
*2.64	Iodonaphthalene	I ⁻		N ₂ /N ₂	111	32
*2.67	Water	(H ₂ O)H ⁺	2.08, 2.13, 2.20, 2.34	O ₂ /Air	160	39
2.68	1-Nitropropane	NO ₂ ⁻		N ₂ /N ₂	108	36
*2.69	Air	(H ₂ O) ₂ O ⁻		Air/N ₂	210	44
*2.69	Air	(H ₂ O) ₂ O ⁻		Air/N ₂	220	43
*2.69	Iodine	I ⁻		N ₂ /N ₂	140	14
*2.70	Air	(H ₂ O) ₂ O ⁻	3.17	Air/SF ₆	210	40
*2.70	m-Mononitrotoluene	(NO ₂) ⁻	1.74	N ₂ /N ₂	166	30
*2.70	m-Mononitrotoluene	(NO ₂) ⁻	1.74, 1.81	Air/N ₂	166	30
*2.70	Air	(NO ₂) ⁻		Air/N ₂	166	41
*2.70	o-Mononitrotoluene	(NO ₂) ⁻	1.74	N ₂ /N ₂	166	30

(Continued on p 170)

TABLE 1 (continued)

K_0	Compound	Ion	Additional product ions	Carrier/drift gases	Temperature ($^{\circ}\text{C}$)	Ref.
*2.70	<i>o</i> -Mononitrotoluene	(NO ₂)	1.74, 1.81	Air/N ₂	166	30
*2.70	<i>p</i> -Mononitrotoluene	(NO ₂) ⁻	1.74	N ₂ /N ₂	166	30
*2.70	<i>p</i> -Mononitrotoluene	(NO ₂)	1.74, 1.81	Air/N ₂	166	30
*2.75	Air	CNO ⁻		Air/N ₂	210	44
*2.75	Air	CNO ⁻		Air/N ₂	220	43
*2.76	Air	CNO ⁻		Air/Air	210	40
2.76	Nitrobutane	(NO ₂) ⁻		N ₂ /N ₂	111	31
*2.76	Air	(NO ₂)		Air/N ₂	210	40
*2.76	Air	(NO ₂)		Air/Air	210	40
*2.77	Air	CNO ⁻		Air/SF ₆	210	40
*2.77	Air	(NO ₂)		Air/SF ₆	210	40
*2.78	Bromine	Br ⁻		N ₂ /N ₂	140	14
2.79	Water	(H ₂ O) ₂ H ⁺	2.27	N ₂ /N ₂	136	15
*2.82	Bromobenzene	Br ⁻		N ₂ /N ₂	111	31
2.90	All 1-chloroalkanes	Cl ⁻		N ₂ /N ₂	135	20
2.90	Chlorobenzene	Cl ⁻		N ₂ /N ₂	136	42
2.91	Chloronitrobenzene	Cl ⁻		N ₂ /N ₂	148	31
2.92	Chlorobenzene	Cl ⁻	1.74	N ₂ /N ₂	140	33
2.92	Chlorobenzene	Cl ⁻		N ₂ /N ₂	125	17
2.92	Decachlorobiphenyl	Cl ⁻	1.08, 1.16	N ₂ /N ₂	125	17
2.92	<i>o</i> -Chloronitrobenzene	Cl ⁻	1.71, 1.86	N ₂ /N ₂	125	17
2.92	Octachlorobiphenyl	Cl ⁻	1.16	N ₂ /N ₂	125	17
2.94	Water	(H ₂ O) ₂ OH ⁻	2.59	Air/N ₂	166	41
2.94	Water	(H ₂ O)OH ⁻	2.59	Air/N ₂	166	41
2.94	Chloride	Cl ⁻		Air/N ₂	166	41
2.95	Dichlorobenzene	Cl ⁻		Air/N ₂	166	41
2.97	Ammonia	NH ₄ ⁺		N ₂ /N ₂	150	5
*3.01	Chloride	-		Air/Air	210	40
*3.01	Chloride	-		Air/SF ₆	210	40
*3.01	Chloride	-		Air/N ₂	210	44
*3.01	Chloride	-		Air/N ₂	220	43
3.08	Ammonium	+		N ₂ /N ₂	200	35
*3.13	Chlorine	-		N ₂ /N ₂	140	14
*3.14	Chloroform	Cl ⁻		N ₂ /N ₂	111	31

*3.17	Air	O ₂ ⁻	2.70	Air/SF ₆	210	40
*3.17	Air	O ₂ ⁻		Air/SF ₆	210	40
*3.17	Air	O ₄ ⁻		Air/SF ₆	210	40
3.28	Oxygen	-		Air/N ₂	166	41
*3.31	Air	-		Air/Air	210	40
*3.31	Air	O ₄ ⁻		Air/Air	210	40
*3.37	Air	CN ⁻		Air/Air	210	40
*3.37	Air	CN ⁻		Air/SF ₆	210	40
*3.37	Air	CN ⁻		Air/N ₂	210	44
*3.37	Air	CN ⁻		Air/N ₂	220	43

where t is the ion's drift time in s, d is the drift length in cm, E is the electric field (V/cm) on the ion drift region, P is the gas pressure in the drift region in Torr and T is the temperature of the drift gas in the drift region in K. In theory, these reduced mobility values are independent of instrumental conditions and accurate drift times can be calculated for any set of instrument parameters. In practice it has been found that often K_0 values do change with instrumental conditions if these conditions are varied over wide ranges. For normal analytical ion mobility detection however, the optimal instrumental conditions do not vary greatly and the reduced mobility values can be used as a reasonably reliable method to predict drift times

2 TABLE OF REDUCED MOBILITY VALUES

We here present a comprehensive compilation of reduced mobility constants from 1970 to 1985 (Table 1) Not only will it provide a means to predict interfering product ions for selective ion mobility detection, it also provides a convenient data base for further investigations of reduced ion mobility constants. A more complete understanding of the behavior of these constants with respect to instrumental parameters is needed in order to produce a truly reliable mobility constant.

The table lists compounds in order of their reduced mobility constants. An asterisk preceding a K_0 value indicates that the mass of that particular ion has been identified with a mass spectrometer. Note that only about a quarter of the entries have been mass identified. For compounds which produce multiple product ions, the additional product ions are listed in the third column of the table, followed by the type of carrier and drift gases used and the temperature employed. Pressure is not reported since only reduced mobility data obtained at atmospheric pressure (760 ± 20 Torr) were considered. The last column provides the reference where more information about the ion mobility spectrum of a particular compound can be found.

While we believe that this table will prove helpful to those who are actively involved in IMS, we would also caution those who use it. Many of the data presented here were obtained in the early days of IMS when the standard practice of sample introduction was to inject vapors or small liquid quantities of the compound directly into the spectrometer. If the compound was contaminated or if the spectrometer became saturated, multiple peaks could occur. In our work where we only introduce nanogram quantities of a compound after separation by high-resolution capillary gas chromatography or supercritical fluid chromatography we have found very few compounds which exhibit multiple product ions.

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4 SUMMARY

This review presents a list of reduced ion mobilities that have been measured under ambient pressure conditions and reported in the open literature during the 16-year period of 1970–1985. Ions reported are listed in order of increasing reduced

mobility along with the name of the parent compound, the reduced mobility of additional product ions observed in the spectrum, the carrier and drift gases, the temperature of the drift region and the reference where the data were reported. Also, ions that have been identified by mass spectrometry are indicated with an asterisk.

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